

REACTION OF ISOCYANIDES. III - SYNTHESIS OF
 β -ALKOXY IMIDOYL CYANIDES FROM ACETALS

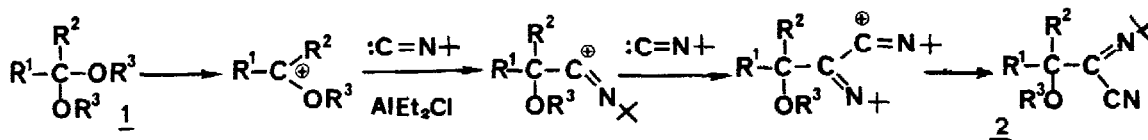
Hélène PELLISSIER and Gérard GIL

Unité Associée au C.N.R.S. n°109 - Faculté des Sciences et Techniques
 Avenue Escadrille Normandie-Niemen - Boîte D12 - 13397 MARSEILLE CEDEX 13

Summary : Tert-butyl isocyanide reacts with acetals, in presence of diethylaluminium chloride, to afford α -iminonitriles.

Litterature reports several syntheses of α -iminonitriles (1-8 and references cited therein). However, few of them give access to imidoynitriles possessing an β -hydrogen or an β heteroatom (9, 10).

Previously, α -alkoxyamides (11), α -alkoxynitriles (12) or β -alkoxycyanoenamines (13) have been synthesized from acetals and isocyanides in the presence of $TiCl_4$. In this paper, we describe a new and easy route to β -alkoxy α -iminonitriles from tert-butyl isocyanide and $AlClEt_2$ (14) as shown in scheme 1.



SCHEME I

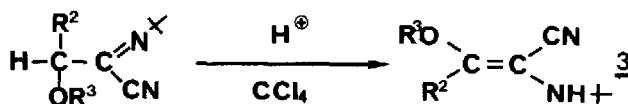
The assignment of the correct structure of 2 is based upon the analysis of NMR, IR and Raman spectroscopic data (14).

For example, in 1H NMR, we observe a singlet at 1.46 ppm corresponding to the tert-butylhydrogens of 2b in full agreement with the results of DE KIMPE (9a) whereas the tert-butylhydrogens of cyanoenamine appear upfield at 1.15 ppm (13).

Using IR spectroscopy, we can identify an absorption at 1640 cm^{-1} ($\gamma_s -C=N-tBu$) but no $C\equiv N$ absorption can be detected. However, an absorption at 2213 cm^{-1} in the Raman spectra of compound 2 points to the presence of the $-C\equiv N$ groups.

Moreover, in ^{13}C NMR a signal at 110 ppm possessing a long relaxation time corresponding to a quaternary carbon can be assigned to the $-C\equiv N$.

When $R^1 = H$, the cyanoimines can undergo an acid-catalyzed isomerization giving the β -alkoxy α -cyanoenamines 3 (Scheme II).



SCHEME II

The experimental results obtained for acetal conversion into iminonitriles 2 are summarized in the following Table.

R ¹	R ²	R ³	%	<u>2</u>	R ¹	R ²	R ³	%	<u>2</u>
CH ₃	H	C ₂ H ₅	90	a	C ₆ H ₅	H	CH ₃	65	e
C ₂ H ₅	H	CH ₃	64	b	(CH ₃) ₂ CH	H	CH ₃	78	f
CH ₃ (CH ₂) ₂	H	CH ₃	63	c	(CH ₃) ₃ C	H	CH ₃	75	g
CH ₃ (CH ₂) ₅	H	CH ₃	90	d	CH ₃	CH ₃	CH ₃	96	h

TABLE

If the use of TiCl₄ or AlClEt₂ allows the preparation of each of the isomeric nitriles 2 and 3, the α -cyanoimines are obtained in better yields than the α -cyanoenamines. The chemical reactivity of the imidoylcyanides 2 is now under investigation in our laboratory.

References :

- 1 - T. SAEGUSA, N. TAKA-ISHI, Y. ITO, *J. Org. Chem.*, **34**, 4040 (1969).
- 2 - J.S. SANDHU, S. MOHAN, A.L. KAPOOR, *Chem. Ind.* (London) **5**, 152 (1971).
- 3 - J.S. SANDHU, S. MOHAN, P.S. SETHI, A.L. KAPOOR, *Indian J. Chem.*, **9**, 504 (1971).
- 4 - M. RAI, K. KRISHAN, A. SINGH, *Indian J. Chem.*, **14B**, 376 (1976).
- 5 - K. TAKAHASHI, *Synthesis*, **1978**, 892.
- 6 - N. DE KIMPE, R. VERHE, L. DE BUYCK, J. CHYS, N. SCHAMP, *Synthesis*, **1978**, 895.
- 7 - F. POCHAT, *Tetrahedron Lett.* **22**, 955 (1981).
- 8 - D.K. DUTTA, D. PRAJAPATI, J.S. SANDHU, J.N. BARUAH, *Synthetic Comm.*, **1985**, 335.
- 9 - a) N. DE KIMPE, R. VERHE, L. DE BUYCK, H. HASMA, N. SCHAMP, *Tetrahedron*, **32**, 3063 (1976).
b) N. DE KIMPE, R. VERHE, L. DE BUYCK, J. CHYS, N. SCHAMP, *Organic Preparations and procedures Int.*, **10**, 149 (1978).
- 10 - N. DE KIMPE, R. VERHE, L. DE BUYCK, J. CHYS, N. SCHAMP, *Bull. Soc. Chim. Belge*, **88**, 695 (1979).
- 11 - T. MUKAIYAMA, K. WATANABE, M. SHIONO, *Chemistry Letters*, **1974**, 1457.
- 12 - Y. ITO, H. IMAI, K. SEGOE, T. SAEGUSA, *Chemistry Letters* **1984**, 937.
- 13 - H. PELLISSIER, A. MEOU, G. GIL, *Tetrahedron Lett.*, **27**, 2979 (1986).
- 14 - The general procedure is described below :

To a stirred solution of tert-butylisocyanide (3 mmol) an acetal (1 mmol) in anhydrous methylene chloride (10 ml) is added dropwise a solution of Et₂AlCl (1.2 mmol), in hexane (10 ml) at room temperature an argon atmosphere. The reaction mixture is stirred 12 h. at room temperature. It is then poured in a saturated aqueous potassium carbonate solution. After usual work-up, the cyanoimine is purified by chromatography on a silica gel column.

2 : R¹ = Et ; R² = H ; R³ = Me : IR (neat) : 2970, 2940, 2880, 2830, 1635 ; ¹H NMR (CDCl₃) : 1.06 (t., 3H) ; 1.46 (s., 9H) ; 1.80 (m., 2H) ; 3.40 (s., 3H) ; 3.78 (t., 1H) ; ¹³C NMR (CDCl₃) : 9.32 (CH₃) ; 25.68 (CH₂) ; 29.32 (CH₃, tBu) ; 57.33 (OCH₃) ; 58.76 (C, tBu) ; 86.97 (OCH) ; 109.83 (C≡N) ; 142.04 (C=N) Anal. : Calcd. for C₁₀H₁₈N₂O : C = 65.89 ; H = 9.95 ; N = 15.37 ; Found : C = 65.23 ; H = 9.62 ; N = 15.42.

(Received in France 20 September 1988)